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You Some Suggested Ones, But You Don't Have To Use

Them. Use What Works For You. SOLUBILITY RULES

(recall That Soluble Means Will Dissolve In Water)

Always Soluble (will Always Completely Dissociate In

Water): Alkali Metal Ions (Li +, Na +, K +, Rb +, Cs +),

NH₄ +, NO₃ ... 4th, 2024.

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... The Formation Of Complex Ions Represents A

Reversible Equilibria Situation. A Complex Ion Is A

Charged Species Consisting Of A Metal Ion Surrounded

By Ligands. A Ligand Is Typically An Anion Or Neutral

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Sparingly Soluble ... Solubility Equilibrium . Solubility Is

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Together, It Is Possible For The Cations From One

Electrolyte To ... Useful To Represent A Metathetical

Reaction By An Ionic Equation, In Which Those Species

That Are ... For The Time Being, We Will Be Using The

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