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Paper 1 Physcical Chemistry Multiple Choice Questions $\Delta H / \text{KJ Mol}^{-1}$ -822.0 0 0 $\Delta \text{MolS} / \text{J K}^{-1}$ -1 90.0 131.0 27.0 The Standard Entropy Value For Steam Is A $+332 \text{ J K}^{-1} \text{ Mol}^{-1}$ B $+189 \text{ J K}^{-1} \text{ Mol}^{-1}$ C $+145 \text{ J K}^{-1} \text{ Mol}^{-1}$ D $+85 \text{ J K}^{-1} \text{ Mol}^{-1}$ (Total 1 Mark) Q25.Which One Of The Equations Below Represents A Reaction That Is Feasible ... Mar 1th, 2024

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