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The Number Of Recorded Counts. This Variation Is Called The Statistical Uncertainty Of A Measurement. It Is An Uncertainty That Is Well Understood. Consider A Measurement That Is Done For A Time Period  $\Delta t$ . Suppose That There Are N Radioactive Nuclei, And That Each One Has A Probability P To Decay In The Time  $3t$ , 2024

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Think Of As Hydrogen, Is Called Hydrogen-1 And Is Stable. The Other Two Isotopes Are Called Hydrogen-2 (deuterium) And Hydrogen-3 (tritium). Deuterium 1th, 2024

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The Binding Energy Per Nucleon Dictates Results When Nuclides Break Apart (fission) Or Fuse Together (fusion) (keep In

Mind That ... (two Protons And Two Neutrons) ! " $= 2 \text{ }^4\text{He} + 2 \dots$  Or An Anti-neutrino (  $\bar{\nu}$  ) Neutrinos Have No Charge, Spin 1/2, And Mass  $\sim 0.1 - 1 \text{ eV}$  (?) The Beta Particle 3th, 2024

### **CHAPTER 1 Radioactive Decay & Decay Modes**

This Equation Represents A First-order Reaction And In The Present Case A Mononuclear Reaction. Radioactive Decay Is Only Possible If  $\Delta E > 0$ .  $\Delta E$  Can Be Determined By Comparison Of The Masses. According To The Relation By Einstein ( $E = Mc^2$ ): (EQ 3) By Calculation Of  $\Delta E$  It C 2th, 2024

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Balancing Decay Reactions: Fill In The Blank In Each Of The Following Decay Reactions With The Correct Decay Particle Or Decayed Nucleus That Will Balance The Reaction, Also State Whether It Is Alpha Or Beta Decay 15.  ${}^{184}_{74}\text{W} \rightarrow {}^{180}_{72}\text{Yb} + {}^4_2\text{He}$  + \_\_\_\_\_ Is An Example Of \_\_\_\_\_ Decay. 16.  ${}^{35}_{16}\text{S} \rightarrow {}^{35}_{17}\text{Cl} + e^{-}$  1th, 2024

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2. Different Isotopes Have The Same Behavior For Chemical Reactions, But React Differently In Nuclear Reactions. 3. Rates Of Nuclear Reactions Are Unaffected By Temperature, Pressure, Catalyst Or Chemical Environment. 4. Energy Changes In Nuclear Reactions Are Million Times Than Chemical 1th, 2024

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