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Mol NaOH) / ... Jul 1th, 2024.

Chapter 17: Acid-Base Equilibria And Solubility

Equilibria) 2SO₄²⁻ That Can Be Added To 150 ML Of

0.050 M BaCl₂ Without Causing A Precipitate To Form?

Solution: First, We Have To Examine A K_{sp} Table (e.g.,

Table 17.4 In The Textbook). We Can Find That The K_{sp}

For BaSO₄ Is 1.1×10^{-10} This Means That If

$[Ba^{2+}][SO_4^{2-}] > K_{sp}$, We Get A Pre Jul 2th, 2024

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