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### **Gas Laws Worksheet (2-08) (modified 3/17) Answer Key**

Gas Laws Worksheet (2-08) (modified 3/17) ... = 14.6 Liter 19.2 Liter X (0.860 g) 25. Think About What Happens When A Bottle Of Carbonated Soft Drink Is Shaken Before Being Opened. Use The Gas Laws To Explain Whether The Effect Will Be Greater When The Liquid Is Warm Or Cold. Shaking The Soda 1th, 2024

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Combined Gas Law T1 P 1 V ... (cm<sup>3</sup>, L, M<sup>3</sup>) Mass (g) Molarity (mol L<sup>-1</sup>) No. Of Particles (particles) No. Of X Atoms (atoms) P P Me Ss 0 23) Y. Kinetic Theory Of Gases Assumptions: 1. A Gas Is Made Up Of Particles Whose Diameters Are Negligible Compared To The ... The Mole And Gas Properties Final 3th, 2024

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Temperature, Pressure, And Volume. Which Gas Has A Greater Number Of Collision Of Gas Molecules With The Walls Of The Container? 1. The He Gas Because It Is Less Massive And Moving With A Higher Average Velocity Correct 2. They Are The Same Since The Pressure Is The Same 3. The O<sub>2</sub> Since It Has A Higher Average Momentum Since It Is More Massive 4. 3th, 2024

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A.) \_\_\_\_\_ Write The Combined Gas Law. B.) \_\_\_\_\_ At What Kelvin Temperature Is There No Kinetic Energy? C.) Match The Graph Below With The Law: Boyle's Or Charles Boyle's Or Charles 2. Solve The Following Problems. Show Your Wo 2th, 2024

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