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23 0.0435 12 0.174 16 0.00726 0.003625 0.003625 0.003625 0.01088 0.003625 Empirical Formula = Na 2CO 3 2 1 3 5) 0.150 G Of Copper Reacts With Oxygen Form 0.188 G Of Copper Oxide. Find The Empirical Formula Of Copper Oxide. Mass Cu = 0.150 G Cu O Mass O = 0.188 - 0.150 = 0.038 G 0.150 63.5 0.038 16 0.00236 0.00236 0.002375 0.00236 Feb 3th, 2024

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C U M F E R E N C E Radius Centre Diameter B A Adj C Opp Hyp Xo W L B B H A H H Compound Measures Speed = Density = Pressure = Volumes Cuboid = L × W × H Prism = Area Of Cross Section × Length Cylinder = π r2h Volume Of Pyramid = × Area Of Base × H Distance Time Mass Volume Force Area Speed Density Pres May 3th, 2024

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