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Calculating PH And POH Worksheet4) -A Solution Is Created By Measuring  $3.60 \times 10^{-3}$  Moles Of NaOH And  $5.95 \times 10^{-4}$  Moles Of HCl Into A Container And Then Water Is Added Until The Final Volume Is 1.00 L. What Is The PH Of This Solution? Since There Is Both Acid And Base We Will Assume A 1 Mole Acid:1 Mole Base Ratio Of Neutralization. 4th, 2024PH And POH - Ms. Mogck's Classroom - Ms. Mogck's ClassroomKEY Chemistry: PH And POH Calculations Part 1: Fill In The Missing Information In The Table Below. PH  $[H^+]$  POH  $[OH^-]$  ACID Or BASE? 3.78  $1.66 \times 10^{-4} M$  10.22  $6.03 \times 10^{-11} M$  Acid 3.41  $3.89 \times 10^{-4} M$  10.59  $2.57 \times 10^{-11} M$  Acid 8.81  $1.55 \times 10^{-9} M$  5.19  $6.46 \times 10^{-6} M$  Base 8.69  $2.04 \times 10^{-9} M$  5.31  $4.88 \times 10^{-6} M$  Base 8.46  $3.47 \times 10^{-9} M$  5.54  $2.88 \times 10^{-6} M$  Base 4th, 2024Basic Concepts Of Chemistry And Chemical CalculationsElements. For Example, Sodium Is A Shiny

Metal, and Chlorine is an irritating gas. But the compound formed from these two elements, sodium chloride, shows different characteristics as it is a crystalline solid, vital for biological functions. Physical Classification Solid Mixtures Ice Water Water Vapour Homogeneous 2th, 2024.

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