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Objective: 1. Observe The Freezing And Melting Behavior Of Lauric Acid 2. Construct A Graph Of The Cooling Curve For Lauric Acid 3. Interpret The Freezing Point And Melting Point From A Graph Background: In Any Pure Substan Apr 1th,

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Represents The Heating Of Substance X At A Constant Rate Of Feb 2th, 2024 Lab #4- Heating Curve - Ms. Hart's Chemistry Class 2013-2014 Versus Temperature Is Called The Heating Curve**. Water Is A Common Substance. Ice Is The Stable Phase Below 0oC. Both Solids And Liquids Coexist At Exactly 0oC. What Does It Mean To Coexist? ____ When Heat Is Put Into The System, More Solid Will Melt. Thus, The Temperature Does Not Change at 0C During The Phase Change. The Normal Boiling ... Jun 2th, 2024.

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Figure 1 Figure 1 shows The Temperature Of 1.00 Kilograms Of Ice. Just Like Heating Curves, Cooling Curves Have Horizontal Flat Parts Where The State Changes From Gas To Liquid, Or From Liquid To Solid. May 1th, 2024.

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