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From Nitrogen And Hydrogen (see Section Chemical  
Equilibrium Of The Thermodynamics Chapter). This  
Reaction Follows The Equation: N<sub>2</sub> + 3H<sub>2</sub> → 2NH<sub>3</sub> (1)  
H<sub>0</sub> = 92 KJ Mol S<sub>0</sub> = 192 J Mol K 4th, 2024Chemical  
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To Flow Reactors With Heat Exchange 542 12.2  
Balance On The Heat-Transfer Fluid 543 12.2.1 Co-  
current Flow 543 12.2.2 Countercurrent Flow 544 12.3  
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1  $A \rightarrow B \rightarrow C$  2.  $A \rightarrow B \rightarrow C$  3.  $A \rightarrow B \rightarrow C$  4.  $A \rightarrow B \rightarrow C$

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