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1. Write The Cation (metal Ion) First And The Anion (nonmetal Ion) Second.
2. Determine The Smallest Whole Number Ratio Of Cations To Anions That Would Make The Charge 0.

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1. Name The Following Ionic Compounds: A. Al_2O_3 Aluminum Oxide B. Cs_2O Cesium Oxide C. Rb_3N Rubidium ... 3th, 2024.

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6-(1-bromoethyl)-5-propylnon-4-ene B. 5-(sec-
butyl)-4-(1-chloropropyl)nona-2,6-diene C.
1-chloro-3-ethyl-2-methylhept-2-ene 4th, 2024.
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Arsenic Trioxide Xenon Hexafluoride Write The Formula For Each Covalent Compound. Boron Trichloride Carbon Dioxide Tetraphosphorus Decoxide Germanium Dichloride Write Two Covalent 4th, 2024 Nomenclature Ultimate Review 2017-Answers Part C — Simple Binary Covalent Compounds I. Provide Names For The Following Covalent Compounds: 1. NF₃ Nitrogen Trifluoride 2. XeO₃ Xenon Trioxide 3. OF₂ Oxygen Difluoride 4. N₂, 2024.

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