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Physical And Chemical Changes Involve Changes In The Potential Energy Of A System, Not The Kinetic Energy. Chemists Use The Term Enthalpy (H) To Refer To The Chemical Potential Energy Of A System, And During Physical And Chemical Changes The System Experiences A Change In Enthalpy (ΔH). 2th, 2024
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Enthalpy Are Denoted By ΔH . ΔH Is Positive When Heat Is Added To The System.
System Surroundings Figure 5.1 The System Is The 3th, 2024Chapter 5
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