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Use The Provided Solubility Graph To Answer The Following ...For Questions 11 - 12, Show Your Work And Circle Your Final Answer. 11. If 115 G  $\text{KNO}_3$  Are Added To 100 G Of Water At  $35^\circ\text{C}$ , How Many Grams Do Not Dissolve? 12. What Mass Of  $\text{KCl}$  Would Be Needed To Form A Saturated Solution If The  $\text{KCl}$  Was Dissolved In 200 G Of Water At  $80^\circ\text{C}$ ? !52 G  $\text{KC}$  1th, 2024Section 7.6: Solubility Equilibria And The Solubility ...Write The Solubility Product Constant Equation. +- #!"!+- -- + + -- ... 2th, 2024Solubility Rules SOLUBILITY RULESSolubility Rules Use Mnemonics To Help You Memorize These!! I'll Give You Some Suggested Ones, But You Don't Have To Use Them. Use What Works For You. SOLUBILITY RULES (recall That Soluble Means Will Dissolve In Water) Always Soluble (will Always Completely Dissociate In Water): Alkali Metal Ions ( $\text{Li}^+$ ,  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Rb}^+$ ,  $\text{Cs}^+$ ),  $\text{NH}_4^+$ ,  $\text{NO}_3^-$  ... 2th, 2024.





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